

Identification of a New Lepidopteran Sex Pheromone in Picogram Quantities using an Antennal Biodetector: (8*E*,10*Z*)-Tetradeca-8,10-dienal from *Cameraria ohridella*

Aleš Svatoš^{*}, Blanka Kalinová, Michal Hoskovec, Jiří Kindl, Oldřich Hovorka,
and Ivan Hrdý

Department of Natural Products, Institute of Organic Chemistry and Biochemistry, Academy of Sciences of the Czech Republic, 166 10 Prague 6, Czech Republic.

Received 22 June 1999; accepted 21 July 1999

Abstract: A major sex attractant released by the virgin female of the horse-chestnut leafminer *Cameraria ohridella* Deschka et Dimić (Lepidoptera: Gracillariidae) which devastates horse-chestnut trees in Europe, was identified in picogram quantities as (8*E*,10*Z*)-tetradeca-8,10-dienal without using spectral methods. The identification solely relied on gas chromatography with electroantennographic detection (GC-EAD), calculation of Kovats' indices of the active principle on different GC phases, and construction of antennal response spectra (EAG response profiles) to C₁₂ and C₁₄ saturated and unsaturated standards with different functional groups. The dienal was prepared by a stereospecific synthesis and shown to be highly active for conspecific males in pg amounts and fully comparable to the natural substance. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: Pheromone; Dienals; Chromatography; Stereocontrol.

Many species of Lepidoptera have developed a remarkably sensitive communication system between sexes whereby males can respond to minute amounts of a pheromone produced by virgin females.¹ Although the structures of Lepidopteran sex pheromones are usually rather simple (straight-chain acetates, alcohols, aldehydes or hydrocarbons),² small amounts of the active compounds, usually found in a complex matrix, make the identification of their chemical structure a challenge. The recent development of spectroscopic techniques (GC/MSⁿ, GC/FTIR) and micro-chemical reactions provides us with methods for the identification of polyunsaturated sex pheromones in nano-gram amounts.³ However, if the amount of the active principle is lower than 10 pg we reach a detection limit of the most sensitive MS instruments (ion traps) and an alternative approach must be sought. Recently, we described chemical communication in the horse-chestnut leafminer *Cameraria ohridella* Deschka et Dimić (Lepidoptera: Gracillariidae).⁴ This pest, which devastates horse chestnut tree foliage, was described⁵ in Macedonia in 1985 and it has colonized most of Europe since that time.⁶ At present, measures against this pest are rather limited (raking of the damaged leaves, spraying with insecticides). The sex pheromone can be a powerful element in designing an integrated pest management (IPM) program to combat this pest.

The preliminary examination of hexane extracts of the calling females using GC-EAD,⁷ where male antennae were used as a biological detector, showed pronounced antennal activity on the EAD trace, but no corresponding GC peak was detected by FID detector (Fig. 1A). Furthermore, when ~ 100 female equivalents (FE) were injected on a GC/MS (ion trap) no reliable mass spectrum was obtained

from the EAD-active area. Clearly, the only analytical tools available were: 1) retention behavior of EAD peak on different GC phases, 2) an examination of antennal specificity to libraries of pheromone-like synthetic compounds (EAG response profiles), and 3) micro-derivatizations of gland extracts combined with EAG.

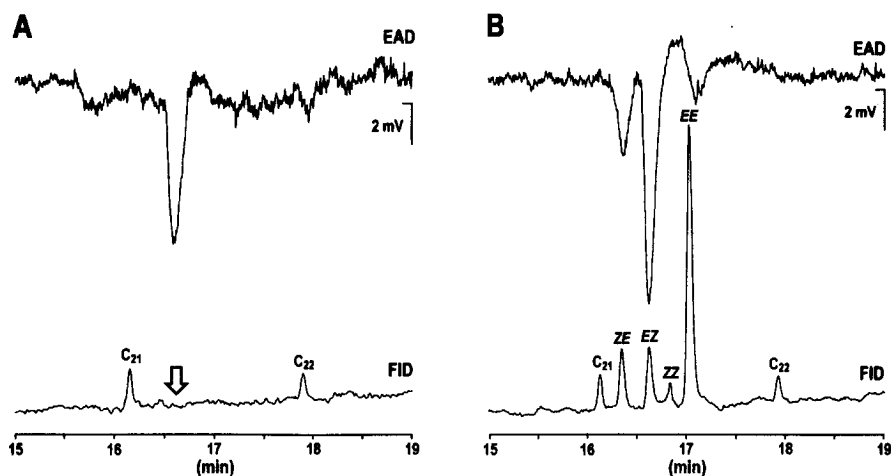


Fig. 1 Sections of GC-FID-EAD traces; **A**: a hexane extract of *C. ohridella* calling females (~ 5 FE); **B**: a synthetic mixture of 8,10-tetradecadienal (7) isomers¹¹ (100 ng) on DB-WAX phase, both co-injected with hydrocarbon standards (C₂₁, 5 ng).

Kovats' indices (KIs) of the EAD peak (Fig. 1A) were determined using several GC phases of increasing polarity and the measured values were compared with KIs of straight-chain aldehydes, alcohols and acetates (Table 1).

Table 1. A comparison of Kovats' indices^a of EAD peak in *C. ohridella* female extracts with some synthetic compounds

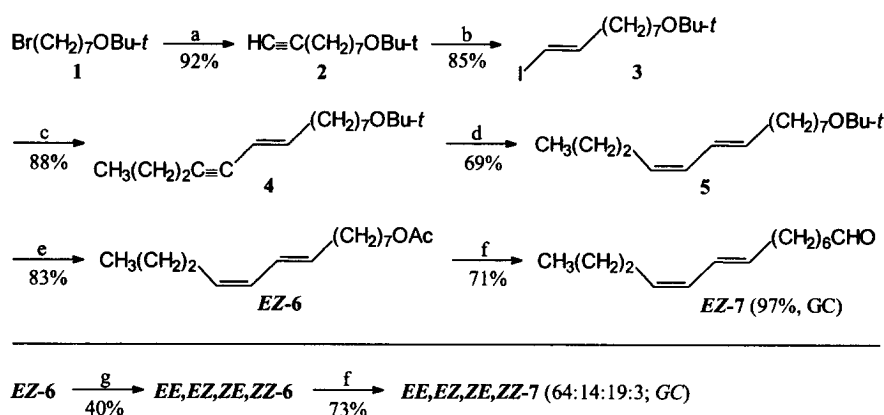
| GC phase ^b | Female extract ^c | Compounds | | | |
|-----------------------|-----------------------------|-----------|--------|--------|-------------------|
| | | 12:Ac | 14:Ald | 14:OH | EZ-7 ^c |
| DB-1 ^d | ca 5 FE | - | - | - | 1623.9 |
| DB-5 ^d | 1674.4 | 1605.9 | 1610.8 | 1675.4 | 1674.3 |
| DB-WAX ^e | 2031.2 | - | - | - | 2031.8 |

^a based on saturated hydrocarbons; ^b J & W Scientific, 30m × 0.25 mm, film thickness 0.25 mm; ^c for EAD trace; ^d 170 °C; ^e 140 - 240 °C @ 5 °C / min

Based on these measurements, a series⁸ of all geometric isomers of dodecen-1-yl acetates, tetradecen-1-ols and tetradecenals and their saturated congeners (1 µg) were tested on the EAG preparation.⁴ Both KIs and the EAG profiles obtained clearly showed that the pheromone should bear the aldehyde functionality and that unsaturation must be situated near the C-9 atom. The aldehydic nature of

the pheromone was confirmed by micro-derivatization experiments where the hexane extracts, treated with *O*-(2,3,4,5,6-pentafluorobenzyl)hydroxylamine hydrochloride in methanol solution,⁹ were tested both on EAG and in a wind tunnel. The pheromonal activity diminished after this derivatization. Unfortunately (*E*)-tetradec-9-enal which had the highest EAG potential (from EAG profiles) showed low behavioral activity (wind tunnel bioassay) and a different KI (1601.8 on DB-5) in comparison to the natural extract. Therefore, it was not considered to be the sex pheromone.

From comparison of KIs on DB-1 and DB-5 phases it seems reasonable to speculate that the pheromone should have more double bonds. Its KI on the DB-5 phase is much higher than the respective KI on a DB-1 phase, which usually points towards conjugation.¹⁰ Based on this consideration, and on the measured EAG profiles, we prepared mixtures of all geometric isomers of 7,9-, 8,10-, and 9,11-tetradecadienals using a general strategy shown for the tetradeca-8,10-dienal (**7**) isomers:



Scheme 1. a) $\text{HC}\equiv\text{CLi}$.EDA / DMSO; b) 1. $\text{AlH}(i\text{-Bu})_2$ / hexane, 2. I_2 / THF; c) 1-pentyne, $\text{Pd}(\text{Ph}_3\text{P})_4$, BuNH_2 , CuI / benzene; d) 1. Dicyclohexylborane, 2. AcOH , 3. $\text{pH } 10$ (NaOH) / H_2O_2 ; e) FeCl_3 / ether / Ac_2O ; f) 1. NaOH / MeOH , 2. PDC / CH_2Cl_2 ; g) PhSH / benzene, 80°C .

The EAG examination of the mixtures of geometric isomers showed that only an isomeric mixture of **7** displays the highest antennal activity. Geometric isomers of the aldehyde **7** were reasonably separated on GC capillary columns and we were able to obtain GC-EAD of the individual isomers.¹¹ Although males' antennae were, to some extent, sensitive to more than one geometric isomer in the mixture we could clearly eliminate the *ZE-7* and *EE-7* isomers. The *EZ-7* isomer showed higher EAD activity than *ZZ-7* isomer (Fig. 1B). When *EZ-7* was measured on GC-EAD using several GC phases the corresponding EAG activity showed identical retention behavior (at sub-ng amounts) to hexane-extracted female abdomens (Table 1). In wind tunnel behavior assay 1 - 0.1 pg of the *EZ-7* isomer displayed high attractiveness, which was comparable to 3 FE of gland extract (100% of all tested males were activated,

took off and 80% of them localized the odor source and tried to copulate with it). In contrast, the pure **ZZ-7** displayed a different KI to the natural extract and its behavioral activity was negligible. In preliminary field experiments sticky traps baited with 5 ng of **EZ-7** isomer (loaded on BBL Taxo paper disc, 1/2 inch dia) were, similarly to virgin females, highly attractive for *C. ohridella* males. All the presented data confirm that **EZ-7** isomer is the main component of *C. ohridella* sex pheromone.

Insect antenna is shown here as a powerful analytical tool when spectroscopic techniques reach their sensitivity limits. It is extremely sensitive to pheromone components *via* specialized pheromone receptors, however, at elevated concentrations it, to a certain extent, responds to pheromone analogs. Consequently, by screening of libraries of the analogs at different concentrations we can direct our structural information towards the pheromone structure.

(8*E*,10*Z*)-Tetradeca-8,10-dienal (**EZ-7**) is, to the best of our knowledge,² a newly discovered sex pheromone and the first identified sex pheromone in genus *Cameraria*. Other isomers of 7 have been described as attractants for males of other Lepidopteran species; **EE-7** for *Acrocercops* sp.¹² and **ZZ-7** for *Phyllonorycter* sp.¹³ The **EZ** conjugated double bond system is quite common, found for example in bombykol, the first described sex pheromone. The high activity of **EZ-7** would be advantageous for establishing of an IPM system based on mating disruption techniques.¹⁴

Acknowledgements: We are highly indebted to Mrs. J. Titzenthalerová and to Mr. O. Blažek for skillful technical assistance in insect handling and to staff of the Royal Garden of the Prague Castle for help in trapping tests. Financial supports from Ministry of the Environment of the Czech Republic and our Institute are acknowledged with pleasure.

References:

1. Kaissling, K. E. *R.H. Wright Lectures on Insect Olfaction*; Simon Fraser University: Burnaby, Canada, 1987; 75 pp.
2. Arn, H.; Toth, M. In *List of Sex Pheromones of Lepidoptera and Related Attractants*; Arn, H.; Tóth, M.; Priesner, E., Eds. The Pherolist. Internet database: www-pherolist.slu.se.
3. Millar, J.G.; Haynes K.F., Eds. *Methods in Chemical Ecology Vol. 1 Chemical Methods*; Kluwer Academic Publishers: Norwell, Massachusetts, USA, 1998; 390 pp.
4. Svatoš, A.; Kalinová, B.; Hoskovec, M.; Kindl, J.; Hrdý, I. *Plant Prot. Science* **1999**, 35, 10-13.
5. Deschka, G.; Dimić N. *Acta Ent. Jugosl.* **1986**, 22, 11-23.
6. Skuhrový, V. *Anzeiger für Schädlingskunde Pflanzenschutz Umweltschutz* **1988**, 71, 82-84.
7. Struble, D.L.; Arn, H. In *Combined Gas Chromatography and Electroantennogram Recording of Insect Olfactory Response*; Hummel, H.; Minks, A. K., Eds. Techniques in Pheromone Research. Springer-Verlag: New York, 1984; pp. 161-178.
8. Alcohols and acetates were available from *PHEROBANK*TM (IPO-DLO, Wageningen, Netherlands), aldehydes were prepared from the corresponding alcohols (PDC / CH₂Cl₂).
9. Svatoš, A.; Urbanová, K.; Boland, W. *in preparation*.
10. Attygalle, A. B.; Morgan, E. D. *Angew. Chem. Int. Ed. Engl.* **1988**, 27, 460-478.
11. Hall, D. R.; Beevor, P. S.; Lester, R.; Poppi R. G.; Nesbitt, B. F. *Experientia* **1980**, 36, 152-154.
12. Ando, T.; Koike, M.; Uchigamu, M., Kuroko, H. *Agric. Biol. Chem.* **1987**, 51, 2691-2695.
13. Reed, D. W.; Chisholm, M. D. *J. Chem. Ecol.* **1985**, 11, 1645-1657.
14. Carde, R. T., Minks, A. K. *Annu. Rev. Entomol.* **1995**, 40, 559-585.